# Electrophilic Substitution in Indoles. Part 19.' Facile Syntheses of the 2a,5a-Diazacyclopenta[j,k]fluorene, Indolo[2,3-a]quinolizinone and Aspidosperma Alkaloid Ring Systems from N-Acyltryptamines 

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#### Abstract

Reaction of tryptamine with diketene gave $N$-[2-(1H-indol-3-yl)ethyl]-3-oxobutyramide (80\%) which with phosphoryl chloride in dichloromethane gave ( $9 \mathrm{~b} S^{*} 9 \mathrm{c} S^{*}$ )-1,2,9b,9c-tetrahydro-5-methyl-2a,5a-diazacyclopenta[j,k]fluoren-3-one 13 (73\%). Hydrogenation gave the 4,5-dihydro and perhydro derivatives. Michael addition of ethyl acetoacetate to benzyl acrylate gave 5-benzyl 1 -ethyl 2-acetylpentanedioate (57\%) which was hydrogenolysed to 4-ethoxycarbonyl-5-oxohexanoic acid $(100 \%)$, the mixed anhydride of which condensed with tryptamine to give 4-ethoxycarbonyl-N-[2-(1H-indol-3-yl)ethyl]-5-oxohexanamide 19 (78\%). The latter, with trifluoracetic acid anhydride gave ( $\pm$ )-cis and trans 1-(ethoxycarbonyl)-2,3,6,7-tetrahydro-12b-methyl-12H-indolo[2,3-a]quinolizin-4(1H)-one (21 and 22) (95\%). N-[2-(1-Methylindol-3-yl)-ethyl]piperidin-2-one 31 was synthesised in three stages. The anion of 31 with diketene gave (a) $N$-[2-(1-methylindol-3-yl)ethyl]-3-(1,3-dioxobutyl)piperidin-2-one, 29 and (b) in a three-stage process, $\quad N$-[2-(1-methylindol-3-yl)ethyl]-3-(1-oxo-2-methoxycarbonylethyl)piperidin-2-one 30. Treatment of the dione 29 with excess of trifluoroacetic acid anhydride gave ( $2 S^{*}, 3 R^{*}, 12 R^{*}$ ) -3-acetyl-5-deethyl-5,19-didehydro-1-methyl-4-oxoaspidospermidine, 34. Reduction of 34 with sodium cyanoborohydride gave the 20,21 -dihydro derivative 35 and two ( $\pm$ )-diastereoisomeric alcohols 36 and 37. Cyclisation of the ester 30 with trifluoracetic acid anhydride gave ( $2 S^{*}, 3 S^{*}, 12 R^{*}$ )-5-deethyl-5,19-didehydro-3-methoxycarbonyl-1-methyl-4-oxoaspidospermidine 39.


In an earlier paper ${ }^{1}$ we showed that cyclisation of $N_{b}$ acetyltryptamines afforded spirocyclic indolines of type 1 in almost quantitative yields. The aim of this work was to develop a synthetic route to derivatives of Aspidosperma indole alkaloids with skeletons of type 2 . Our intention was to use the nucleophilic activity ${ }^{1}$ at the indole 3-position of suitably elaborated acyl tryptamines for closure of the 12,19-ring junction, and to complete the pentacyclic system by forming the 2,3-bond via attack of a nucleophilic centre at C-2


1


4



3

Since the pioneering synthesis of strychnine by Woodward et al., ${ }^{2}$ this type of approach has been exploited by several groups. For example by Van Tamelen et al., ${ }^{3}$ Buchi et al., in their synthesis of vindorosine, ${ }^{4}$ Pandit et al. ${ }^{5}$ in their synthesis of the aspidosperma skeleton and Wenkert et al. ${ }^{6}$ who described a synthesis of deethylvincadifformine 3 in which the key
step was the boron trifluoride-catalysed cyclisation of the acyl enamine 4. Introduction of an ethyl group into the starting pyridine derivative led to the synthesis of the natural compounds. ${ }^{7}$
In the present work, we began with attempts to cyclise relatively simple amides such as 5 with potential nucleophilic character at the C-3 position. Acylation, ${ }^{8}$ like other electrophilic reactions, ${ }^{9}$ proceeds at the indole 3-position, hence we expected that amides such as 5 under Bischler-Napieralski conditions would lead initially to the spirocyclic indolenium intermediate 6 (Scheme 1). Whilst normally, rearrangement of 6 to a dihydro- $\beta$-carboline would occur, in our case we hoped that the potentially nucleophilic centre at C-3 would attack the indoleninium function to give the pyrrolo $[2,3-d]$ carbazole system 7. For a successful functional group (X), more elaborate amides could be envisaged such as 8 which, on cyclisation,



might afford the pentacyclic skeleton 9. Our proposed route differed from previous ones in that the starting amides were structurally simpler and in a different oxidation state from the various enamine derivatives referred to above. Preliminary accounts of some of this work have been reported. ${ }^{10,11}$
We first decided to use the keto amide 10 which might cyclise to the tetracyclic indoline 11 (Scheme 2). Reaction of


Scheme 2
tryptamine with diketene gave an $80 \%$ yield of the required amide 10. Attempted cyclisations with trifluoroacetic acid anhydride or boron trifluoride dietherate were unsuccessful, but phosphoryl chloride in dichloromethane at $20^{\circ} \mathrm{C}$ for 24 h , gave the indoline $13(73 \%)^{*}$ after chromatography, as an oil, $\mathrm{M}^{+} 226.1106\left(100 \%, \mathrm{C}_{14} \mathrm{H}_{14} \mathrm{~N}_{2} \mathrm{O}\right)$.

In the ${ }^{1} \mathrm{H}$ NMR spectrum a doublet at $\delta 5.5(J 8 \mathrm{~Hz})$ could be assigned to the $9 \mathrm{c}-\mathrm{H}$ which correlated well with other tetrahydropyrroloindoles. ${ }^{12,13}$ The four aromatic protons appeared at higher field than $\delta 7.27$ in keeping with the change from indole to indoline; the remaining signals were assigned from decoupling experiments. The ${ }^{13} \mathrm{C}$ NMR spectrum was assigned by comparison with the literature, ${ }^{7}$ in particular $\mathrm{C}-9 \mathrm{~b}$ and $\mathrm{C}-9 \mathrm{c}$ gave two doublet signals at $\delta 45.1$ and 79.0. This cyclisation may involve activation of the acetyl group (Scheme 2) in which prior attack of phosphoryl chloride on the ketone with release of HCl is a controlling feature. The indoline 13 was

[^0]unstable, reverting to starting material when stored for a few days. Hydrogenation at atmospheric pressure gave the stable amine 15 and the perhydro derivative 16.


15


15a


16


In the ${ }^{1} \mathrm{H}$ NMR spectrum of 15 the methyl group ( $3 \mathrm{H}, \mathrm{d}, \mathrm{J}$ 7 Hz ) at $\delta 1.6$ was shown to be trans to the $9 \mathrm{~b} / 9 \mathrm{c}-\mathrm{H}$ atoms by NOE difference spectra (see 13a). For the fully hydrogenated derivative 16 the NMR assignments and coupling constants for most of the protons were deduced from a 2D COSY 45 experiment. NOE difference spectra gave the enhancements shown in 16a. The stereostructure of the $5 \mathrm{~b}-9 \mathrm{a}$ ring junction was not determined, but presumably hydrogen was delivered catalytically to the same face of the molecule as for the saturation of the C-4 C-5 double bond of 13, and therefore the cis-cis structure 16a would be expected.

The 5-methoxy analogue 11 of the starting amide was also synthesised from 5-methoxytryptamine and when treated with phosphoryl chloride under similar conditions to those used for 10, 11 gave the 8 -methoxyindoline $14(68 \%)$. The diazacyclopenta $[j, k] f l u o r e n e ~ r i n g ~ s y s t e m ~ o f ~ 13 ~ a n d ~ 14 ~ h a s ~ b e e n ~ r e p o r t e d ~$ previously only as a product of cyclisation of $N$-substituted indoles. ${ }^{14}$ The approach via the amides 10 and 11 described above is a novel although unexpected route.
We then moved to a more elaborate amide precursor. Michael addition of ethyl acetoacetate to benzyl acrylate gave the benzyl ester $17(57 \%)$ which was hydrogenolysed to the corresponding acid $18(100 \%)$ over palladised charcoal. Condensation of the acid with tryptamine using dicyclohexylcarbodiimide in tetrahydrofuran afforded only a small amount of product, but treatment of the acid 18 with 1 equiv. each of ethyl chloroformate and triethylamine at $-5^{\circ} \mathrm{C}$ gave the mixed anhydride which, without isolation, was condensed with tryptamine to give the amide 19 as a brown, viscous oil ( $78 \%$ ) after chromatography. Similarly the methoxy substituted analogue $\mathbf{2 0}$ was obtained in $32 \%$ yield.

Treatment of the amide 19 with 1 equiv. of trifluoracetic acid anhydride in benzene gave, as the main product, the indolo[2,3a]quinolizinone 21. In acetonitrile as solvent, the cis-21 and trans-22 isomers were isolated after chromatography in the ratio of $5: 1$ respectively, whereas with phosphoryl chloride ( 1 equiv.) as the cyclisation reagent the cis isomer 21 was obtained in a $99: 1$ excess and in $>95 \%$ yield. In the ${ }^{1} \mathrm{H}$ NMR spectrum of 21, the indole-2-H was absent and there was a series of multiplets in the region $\delta 2-3$, assigned by decoupling experiments to the 2,3,6 and 7 protons. In particular the $6 \alpha-\mathrm{H}$ proton at $\delta 5.07$, in the deshielding region of the amide carbonyl, showed the expected lowfield shift observed in model compounds. ${ }^{15}$ The signal, a double double doublet, $J 13,4,2$ Hz , is acceptable for the part structure/conformation 21a.

For the trans isomer 22, the $\mathrm{CH}_{2} \mathrm{CH}_{3}$ signals of the ester were considerably shielded compared with the cis isomer 21 as

$17 \mathrm{R}=\mathrm{CH}_{2} \mathrm{Ph}$ $18 \mathrm{R}=\mathrm{H}$

$19 R=H, R^{\prime}=H$
$20 R=O M e, R^{\prime}=H$

$211-\beta \mathrm{CO}_{2} \mathrm{Et}, \mathrm{R}=\mathrm{H}$
$221-\alpha \mathrm{CO}_{2} E \mathrm{Et}, \mathrm{R}=\mathrm{H}$ $231-\beta \mathrm{CO}_{2} \mathrm{Et}, \mathrm{R}=\mathrm{OMe}$


22a

$24 \alpha-\mathrm{CO}_{2} \mathrm{Me}$
$25 \mathrm{~B}-\mathrm{CO}_{2} \mathrm{Me}$
was the indole-NH signal. These effects can be explained by models which show that, only in the trans isomer 22, with an axial ethoxycarbonyl group, the $\mathrm{CH}_{2} \mathrm{CH}_{3}$ protons are shielded by the indole ring system for a significant part of the rotation about the axial bond. A similar effect has been reported ${ }^{16}$ for the isomers 24 and 25 . The signals in rings C and D of the trans isomer 22 were assigned by decoupling experiments and the part-structure/conformation 22a was supported by the NOE enhancements shown.

In the reaction of the amide 19 with only 1 equiv. of trifluoroacetic acid anhydride, activation by the anhydride is presumably selective at the keto group, which results in attack at this point by the amide nitrogen (Scheme 3). In the resultant iminium species 26 nucleophilic attack by the indolyl-3-position would be expected to occur predominantly on the opposite side to the bulky ethoxycarbonyl group leading mainly to the cis isomer 21 observed.

The 5-methoxy derivative 20 of the amide, when treated with 1 equiv. of phosphoryl chloride in acetonitrile, gave the cis 9 -methoxyindolo[2,3-a]quinolizinone 23 in $63 \%$ yield. The indolo [2,3-a]quinolizinone system with a 12 b methyl group was first reported by Winterfeld ${ }^{17}$ in 1964.

Following these results with the amides 19 and 20 we next used 10 equiv. of trifluoroacetic acid anhydride as the cyclisation reagent for 19 in dry benzene at $20^{\circ} \mathrm{C}$. After 72 h , a complex mixture was obtained. Chromatography gave an impure, unstable, yellow crystalline product tentatively identified as the pyrrolo[2,3-d]carbazole 27 ( $50 \%$ ), which showed $\mathrm{M}^{+}$at $m / z 420(100 \%)$ in its FD mass spectrum. The


Scheme 3


27


28

$29 \mathrm{R}=\mathrm{Me}$
$30 \mathrm{R}=\mathrm{OMe}$
${ }^{1} \mathrm{H}$ NMR spectrum indicated that this product was mainly one component and the absence of the indole- 2 H signal and the 1 ppm shift to lower field of the acetyl signal, compared with the amide 19, were in keeping with the structure proposed. The four multiplets at $\delta 2.5,2.84,4.86$ and 5.44 were shown to be coupled by double resonance experiments and were assigned by comparison with the imino-ketone $28 .{ }^{18}$ The signals for the $12-\mathrm{H}(\delta 8.25$, d) and $13 \mathrm{a}-\mathrm{H}(\delta 6.44$ and $6.32,2 \times \mathrm{s})$ both appear at lower field than normally expected due to deshielding by the $N$-trifluoroacetyl group and the latter proton appears as two signals, probably due to hindered rotation about the amide bond. The remaining aromatic signals and the $1-\mathrm{H}$ and $4-\mathrm{H}$ signals were unexceptional and assigned by spindecoupling experiments.

Unfortunately, although the cyclisation reaction was repeated a number of times the proposed imine 27 could not be isolated although it was identified in some of the mixtures of products. Although we did not obtain it as a completely pure product, its transient formation suggested that we should persist with this approach. We therefore moved to the synthesis and cyclisation of the lactams 29 and 30 ; in these cases both cyclisation (as for 13) on the indole nitrogen and condensation between amide nitrogen and ketone (as for 21) would be precluded. Spirocyclic ring formation and anionic attack as outlined in Scheme 1 should be more successful than for the amides 19 and 20.

The lactam 31 was synthesised as shown in Scheme 4 from 5 -bromopentanoic acid ${ }^{19}$ in $71 \%$ yield. Treatment of this lactam 31 with lithium diisopropylamide ( 2 equiv.) and diketene ( 1 equiv.) gave the dione 29 as an oily mixture of the predominantly two enol and one keto forms 29a, b, c ( $35 \%$ ). The two enol protons were evident from signals at $\delta 15.20$ and


Scheme 4 Reagents: i, oxalyl chloride; ii, tryptamine; iii, KH ; iv, NaH ; v, MeI

15.36. The lithium enolate of the lactam 31 on treatment with benzyl chloroformate gave the benzyl ester 32 ( $73 \%$ ), the anion of which was acylated with methoxycarbonylacetyl chloride to give the diester 33 in $48 \%$ yield. Conventional hydrogenolysis of $\mathbf{3 3}$ proved unsuccessful but the use of ammonium formate with $10 \%$ palladised charcoal ${ }^{20}$ gave the keto ester 30 ( $64 \%$ ) directly as a mixture of keto and enol forms $30 \mathbf{a}$ and $\mathbf{b}$.
Treatment of the dione 29 with 10 equiv. of trifluoroacetic acid anhydride for 24 h , gave the pentacyclic amide 34 as the only isolated product ( $60 \%$ ). The amide, $\mathrm{M}^{+} 322$, showed $v_{\text {max }} 1710$ and $1645 \mathrm{~cm}^{-1}$, and $\lambda_{\text {max }} 253$ and 308 nm , characteristic of a vinylogous amide chromophore. In the ${ }^{1} \mathrm{H}$ NMR spectrum the 2-H signal was a doublet, $J 8 \mathrm{~Hz}$ at $\delta 4.44$, shown by double resonance to be coupled to the doublet at $J$ $3.40(3-\mathrm{H})$ in keeping with a trans diaxial configuration. The NOE enhancements shown, the ${ }^{13} \mathrm{C}$ NMR spectrum and elemental analysis confirmed the structure of 34.
Hydrogenation of the tetrasubstituted double bond of 34 predictably, proved impossible with platinum oxide catalyst with or without ammonium formate. Reduction with sodium cyanoborohydride ${ }^{21}$ however, gave three products. The least polar on chromatography was the diketone $35, \mathrm{M}^{+} 324$, which in the ${ }^{1} \mathrm{H}$ NMR spectrum solution existed entirely in the enol


34


35
form shown. The enol-H and 2-H signals were singlets at $\delta$ 16.27 and 3.88 respectively. A doublet at $\delta 2.22(J 3 \mathrm{~Hz})$ was assigned to the 19-H proton. Closely analogous cis $\mathrm{C}_{5}-\mathrm{C}_{19}$ junction compounds show doublets with $J 3^{22}$ and $4 \mathrm{~Hz}^{7}$ for the $19-\mathrm{H}$ whereas the trans ring junction has $J_{5,19} \approx 12 \mathrm{~Hz}$. The UV spectrum also [ $\lambda_{\text {max }} 256$ ( 816 200) and 284 (11770)] supported the enone chromophore.
The remaining two products were the diastereoisomeric alcohols 36 and 37 which still showed the vinylogous amide

chromophone ( $\lambda_{\text {max }} 308 \mathrm{~nm}$ ). Both isomers showed a multiplet at $\delta 4.10$ in their ${ }^{1} \mathrm{H}$ NMR spectra. In the more polar isomer this was shown by double resonance to be coupled to the 3-H (signal at $\delta 2.30$, dd, $J 6.5,3 \mathrm{~Hz}$ ) and to the methyl protons ( $\delta$ 1.38 ) and in the less polar isomer to corresponding signals at $\delta$ 2.45 and 1.20 respectively. The $2-\mathrm{H}$ signal in each isomer was a doublet, $J 6.5 \mathrm{~Hz}$ at $\delta 3.86$ and 3.45 respectively. These alcohols were not investigated further.
Attempts were made to cyclise the dione 29 with phosphoryl chloride and boron trifluoride-diethyl ether. With the former, no reaction took place even with 10 equiv. of reagent after 20 h . With the latter, only the boron complex 38 was isolated $(38 \%)$. The EI mass spectrum gave $\mathrm{M}^{+}(9 \%)$ at 388.1754 $\left(\mathrm{C}_{20} \mathrm{H}_{23} \mathrm{BF}_{2} \mathrm{~N}_{2} \mathrm{O}_{3}\right)$. The ${ }^{1} \mathrm{H}$ NMR spectrum showed the indole-2-H proton as a singlet, $\delta 6.96$ and 2 and 3 proton singlets at $\delta 3.43$ and 2.28 for the terminal part of the butanedione group, and this, with the normal $\mathrm{C}=\mathrm{O}$ bond at $1720 \mathrm{~cm}^{-1}$ in the IR indicated the position of the boron shown.
Finally, the $\beta$-keto ester 30 was treated with 10 equiv. of trifluoroacetic acid anhydride for 24 h , to give the pentacyclic ester 39 as a colourless crystalline solid ( $51 \%$ ). The spectral properties of the ester were very similar to those of the 16 acetyl analogue 34; the 2-H and 3-H signals were the expected doublets ( $J 8 \mathrm{~Hz}$ ) at $\delta 4.32$ and 3.24 respectively.
These results demonstrate that cyclisation of relatively simple tryptamides under different conditions can give rise to a variety of polycyclic heterocycles in good yields.

## Experimental

IR spectra were recorded on a Unicam SP200 grating spectrophotometer, UV spectra were measured in absolute ethanol on a Unicam SP800 spectrophotometer. ${ }^{1} \mathrm{H}$ NMR spectra were recorded on a Bruker WM 360 ( 360 MHz ) spectrometer in $\mathrm{CDCl}_{3}$ unless stated otherwise. ${ }^{13} \mathrm{C}$ NMR spectra were recorded on a Bruker WM 360 spectrometer at 90.56 MHz ; $J$ values are given in Hz . Mass spectra were electron impact (EI), field desorption (FD) or chemical ionisation (CI, using either ammonia or isobutane as the reagent) and were measured on a Varian CH 5D double focussing instrument.

Flash chromatography was carried out as described by Still et al. ${ }^{23}$ using silica gel S (Merck) 230-400 mesh. Light petroleum refers to a fraction of boiling range $40-60^{\circ} \mathrm{C}$. Dimethylformamide and diisopropylamine were distilled from calcium hydride under nitrogen before use.

N-[2-(1H-Indol-3-yl)ethyl]-3-oxobutyramide 10.--Tryptamine ( $1 \mathrm{~g}, 6.25 \mathrm{mmol}$ ) in tetrahydrofuran (THF) $\left(30 \mathrm{~cm}^{3}\right)$ was stirred at $0^{\circ} \mathrm{C}$ under nitrogen. Diketene ( $0.53 \mathrm{~g}, 6.25 \mathrm{mmol}$ ) was added over 15 min . The reaction mixture was warmed to $20^{\circ} \mathrm{C}$ and stirred for 2 h , quenched with saturated aqueous sodium hydrogen carbonate and then extracted with dichloromethane ( $3 \times 30 \mathrm{~cm}^{3}$ ). The dichloromethane extracts were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and then the solvent was removed to give a red-brown oil which crystallised from chloroform-diethyl ether to give the amide 10 as colourless needles ( $1.23 \mathrm{~g}, 80 \%$ ); m.p. $77-78{ }^{\circ} \mathrm{C}$; $\delta_{\mathrm{H}} 2.22\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3} \mathrm{CO}\right), 3.0(2 \mathrm{H}, \mathrm{t}, J 7$, Ind$\left.\mathrm{CH}_{2}\right), 3.34\left(2 \mathrm{H}, \mathrm{s}, \mathrm{COCH}_{2} \mathrm{CO}\right), 3.63(2 \mathrm{H}, \mathrm{q}, J 7$, Ind$\left.\mathrm{CH}_{2} \mathrm{CH}_{2}\right), 6.93(1 \mathrm{H}, \mathrm{br}$ s, NHCO), $7.06(1 \mathrm{H}, \mathrm{s}$, Ind-2-H), 7.12 ( $1 \mathrm{H}, \mathrm{t}, J$, Ind-5-H), $7.21(1 \mathrm{H}, \mathrm{t}, J$ 8, Ind-6-H), $7.37(1 \mathrm{H}, \mathrm{d}$, $J 8$, Ind-7-H), $7.62(1 \mathrm{H}, \mathrm{d}, J 8$, Ind-4-H) and $8.15(1 \mathrm{H}, \mathrm{br} \mathrm{s}$, Ind-NH); $v_{\max }\left(\mathrm{CHCl}_{3}\right) / \mathrm{cm}^{-1} 1712$ and $1665 ; \lambda_{\text {max }} / \mathrm{nm} 223.5$ $\left(\varepsilon / \mathrm{dm}^{3} \mathrm{~mol}^{-1} \mathrm{~cm}^{-1} 21350\right) 274.5$ (3690) 282 (3870) and 291 (3290); m/z (EI) 244 (58\%) 143 (98) and 130 (100); (FD) 244 (100) (Found: $\mathrm{M}^{+}$244.1211. $\mathrm{C}_{14} \mathrm{H}_{16} \mathrm{~N}_{2} \mathrm{O}_{2}$ requires $M$, 244.1211).
( $9 \mathrm{bS} \mathbf{S}^{*}, 9 \mathrm{cS}{ }^{*}$ )-1,2,9b,9c-Tetrahydro-5-methyl-2a,5a-diazacyclopenta[j, k$]$ fluoren-3-one 13.-The amide $10(0.5 \mathrm{~g}, 2.05 \mathrm{mmol})$ in dichloromethane ( $20 \mathrm{~cm}^{3}$ ) was stirred at $20^{\circ} \mathrm{C}$ under an atmosphere of nitrogen. Phosphoryl chloride $(0.314 \mathrm{~g}, 1.1$ equiv.) was added over 5 min and the solution was stirred for 24 h , at $20^{\circ} \mathrm{C}$. The dark green solution was then washed with dilute aqueous ammonia and the aqueous layer extracted with dichloromethane ( $2 \times 20 \mathrm{~cm}^{3}$ ). The combined organic layers were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and then the solvent was removed under reduced pressure to give a red-brown oil which was purified by flash chromatography. Elution with $2 \%$ methanol in ethyl acetate gave the indoline 13 as a red oil $(0.34 \mathrm{~g}, 73 \%) ; \delta_{\mathrm{H}} 1.9$ $(1 \mathrm{H}, \mathrm{m}, 1-\mathrm{H}), 2.35\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right), 2.45(1 \mathrm{H}, \mathrm{m}, \mathrm{l}-\mathrm{H}), 3.4(1 \mathrm{H}$, $\mathrm{m}, 2-\mathrm{H}), 3.8(1 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}), 4.0(1 \mathrm{H}, \mathrm{m}, 9 \mathrm{~b}-\mathrm{H}), 5.25(1 \mathrm{H}, \mathrm{s}$, $4-\mathrm{H}), 5.5(1 \mathrm{H}, \mathrm{d}, J 8,9 \mathrm{c}-\mathrm{H}), 6.92(1 \mathrm{H}, \mathrm{t}, J 8,8-\mathrm{H}), 6.98(1 \mathrm{H}, \mathrm{d}$, $J 8,9-\mathrm{H}), 7.16(1 \mathrm{H}, \mathrm{t}, J 8,7-\mathrm{H})$ and $7.23(1 \mathrm{H}, \mathrm{d}, J 8,6-\mathrm{H}) ; \delta_{\mathrm{c}}$ 20.7, 31.1, 44.0, 45.1, 79.0, 103.7, 111.6, 121.8, 125.3, 128.6, 134.2, 143.7, 149.4 and 163.1; $v_{\max }\left(\mathrm{CHCl}_{3}\right) / \mathrm{cm}^{-1} 1630$ and 1605; $\lambda_{\text {max }} / \mathrm{nm} 230\left(\varepsilon / \mathrm{dm}^{3} \mathrm{~mol}^{-1} \mathrm{~cm}^{-1} 6120\right)$ and 326.5 (6240); $m / z$ (EI) 226 ( $\mathrm{M}^{+}, 100 \%$ ), 225 (67), 198 (13), 184 (10), 170 (20), 144 (10) and 130 (15); (FD) 226 (100) (Found: M $^{+} 226.1106$. $\mathrm{C}_{14} \mathrm{H}_{14} \mathrm{~N}_{2} \mathrm{O}$ requires $M, 226.1106$ ).

Hydrogenation of the Indoline 13.--The indoline 13 ( 98 mg , 0.44 mmol ) in ethanol ( $3 \mathrm{~cm}^{3}$ ) was hydrogenated at atmospheric pressure using platinum oxide catalyst. When the uptake of hydrogen ceased ( 30 min ) the catalyst was removed by filtration and then the solvent was removed under reduced pressure to give a brown solid which was purified by flash
chromatography. Elution with $1 \%$ methanol in chloroform gave the ( $\pm$ )-indoline 15 as a colourless solid ( $59 \mathrm{mg}, 59 \%$ ); m.p. $142{ }^{\circ} \mathrm{C}$ (methanol); $\delta_{\mathrm{H}} 1.6\left(3 \mathrm{H}, \mathrm{d}, J 7, \mathrm{CH}_{3}\right), 2.15(1 \mathrm{H}, \mathrm{dd}$, $J 17,5,4-\mathrm{H}), 3.3(1 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}), 3.75(1 \mathrm{H}, \mathrm{m}, 9 \mathrm{~b}-\mathrm{H}), 3.85(1 \mathrm{H}$, $\mathrm{m}, 2-\mathrm{H}), 4.0(1 \mathrm{H}, \mathrm{m}, 5-\mathrm{H}), 5.15(1 \mathrm{H}, \mathrm{d}, J 6.4,9 \mathrm{c}-\mathrm{H}), 6.84$ ( $2 \mathrm{H}, \mathrm{m}, \mathrm{ArH}$ ) and $7.1(2 \mathrm{H}, \mathrm{m}, \mathrm{ArH}) ; \delta_{\mathrm{c}} 20.0,28.2,36.7,43.1$, 45.2, 50.2, 81.6, 111.7, 120.1, 125.0, 128.0, 133.6 and 167.6; $v_{\text {max }}\left(\mathrm{CHCl}_{3}\right) / \mathrm{cm}^{-1} 1635 ; \lambda_{\text {max }} / \mathrm{nm} 244\left(\varepsilon / \mathrm{dm}^{3} \mathrm{~mol}^{-1} \mathrm{~cm}^{-1} 8770\right)$ and 292 (2820); $m / z$ (EI) $228\left(\mathrm{M}^{+}, 70 \%\right.$ ) 213 (100) 170 (60) and 130 (20) (Found: $\mathrm{M}^{+} 228.1258 . \mathrm{C}_{14} \mathrm{H}_{16} \mathrm{~N}_{2} \mathrm{O}$ requires $M$, 228.1263).

Further elution gave the ( $\pm$ )-perhydro compound 16 as colourless crystals ( $15 \mathrm{mg}, 15 \%$ ); m.p. $102-104^{\circ} \mathrm{C} ; \delta_{\mathrm{H}} 1.2-1.9$ $\left(9 \mathrm{H}, \mathrm{m}, 6-, 7-, 8-\right.$ and $9-\mathrm{H}_{2}$ and $\left.1 \alpha-\mathrm{H}\right), 1.35\left(3 \mathrm{H}, \mathrm{d}, J 7, \mathrm{CH}_{3}\right)$, $2.1(2 \mathrm{H}, \mathrm{m}, 1 \beta-\mathrm{H}, 4 \alpha-\mathrm{H}), 2.4(2 \mathrm{H}, \mathrm{m}, 9 \mathrm{a}-\mathrm{H}$ and $4 \beta-\mathrm{H}), 2.9$ ( $2 \mathrm{H}, \mathrm{m}, 9 \mathrm{~b}-\mathrm{H}$ and $2 \alpha-\mathrm{H}$ ), $3.2(1 \mathrm{H}, \mathrm{m}, 6 \mathrm{a}-\mathrm{H}), 3.55(1 \mathrm{H}, \mathrm{m}$, $5 \beta-\mathrm{H}), 4.33(1 \mathrm{H}$, ddd, $J 14,10,2,2 \beta-\mathrm{H})$ and $4.70(1 \mathrm{H}, \mathrm{d}, J 7$, $9 \mathrm{c}-\mathrm{H}) ; v_{\max }\left(\mathrm{CHCl}_{3}\right) / \mathrm{cm}^{-1} 1625 ; m / z(\mathrm{EI}) 234\left(\mathrm{M}^{+}, 25 \%\right) 219$ (30) and 191 (100) (Found: $\mathrm{M}^{+}$234.1730. $\mathrm{C}_{14} \mathrm{H}_{22} \mathrm{~N}_{2} \mathrm{O}$ requires $M, 234.1732$ ).

N-[2-(5-Methoxy-1H-indol-3-yl)ethyl]-3-oxobutyramide 11.-To a solution of 5-methoxytryptamine ( $0.49 \mathrm{~g}, 2.6 \mathrm{mmol}$ ) in tetrahydrofuran ( $15 \mathrm{~cm}^{3}$ ) under an atmosphere of nitrogen at $-5^{\circ} \mathrm{C}$ was added diketene ( $0.22 \mathrm{~cm}^{3}, 2.9 \mathrm{mmol}$ ) dropwise, over 10 min . The solution was allowed to warm to $20^{\circ} \mathrm{C}$ and stirred for 24 h , by which time TLC analysis showed the reaction to be complete. The solvent was removed under reduced pressure to give a brown oil ( 0.87 g ) which on flash chromatography and elution with chloroform-methanol ( $98: 2$ ) gave the amide 11 as a brown oil $(0.71 \mathrm{~g}, 99 \%) ; \delta_{\mathrm{H}} 2.22(3 \mathrm{H}, \mathrm{s}$, $\mathrm{CH}_{3} \mathrm{CO}$ ), $2.96\left(2 \mathrm{H}, \mathrm{t}, J 7\right.$, Ind- $\mathrm{CH}_{2}$ ), $3.36\left(2 \mathrm{H}, \mathrm{s}, \mathrm{COCH}_{2} \mathrm{CO}\right.$ ), $3.61\left(2 \mathrm{H}, \mathrm{m}, \mathrm{Ind}-\mathrm{CH}_{2} \mathrm{CH}_{2}\right), 3.88(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}) 6.85(1 \mathrm{H}, \mathrm{dd}$, $J$ 8, 3, Ind-6-H), $6.95(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{NHCO}) 7.05(2 \mathrm{H}, \mathrm{d}, J$ 3, Ind-4-H and Ind-2-H), 7.25 ( $1 \mathrm{H}, \mathrm{d}, J 8$, Ind-7-H) and $8.1(1 \mathrm{H}, \mathrm{br}$ s , Ind-NH); $v_{\text {max }}$ (liq. film)/ $\mathrm{cm}^{-1} 3340,1715$ and 1650 ; $\lambda_{\text {max }} / \mathrm{nm} 226\left(\varepsilon / \mathrm{dm}^{3} \mathrm{~mol}^{-1} \mathrm{~cm}^{-1} 36990\right), 278$ (8320) and 297.5 (6680); m/z (EI) 274 (M ${ }^{+}, 2 \%$ ), 173 (50) and 160 (100); (FD) 274 (100) (Found: $\mathrm{M}^{+}$274.1321. $\mathrm{C}_{15} \mathrm{H}_{18} \mathrm{~N}_{2} \mathrm{O}_{3}$ requires $M$, 274.1317).
(9bS*,9cS*)-1,2,9b,9c-Tetrahydro-8-methoxy-5-methyl-2a,5adiazacyclopenta[j, k$]$ fluoren-3-one 14 .-To a solution of the amide $11(0.2 \mathrm{~g}, 0.73 \mathrm{mmol})$ in dichloromethane $\left(15 \mathrm{~cm}^{3}\right)$ at $20^{\circ} \mathrm{C}$ under an atmosphere of nitrogen, was added phosphoryl chloride ( $74 \mathrm{~mm}^{3}, 0.8 \mathrm{mmol}$ ). After being stirred for 6 h , TLC showed the reaction to be complete. The solution was washed with dilute aqueous ammonia, the aqueous layer extracted with dichloromethane ( $2 \times 20 \mathrm{~cm}^{3}$ ) and the combined organic layers dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$. The solvent was removed under reduced pressure to give a pale yellow oil ( 0.16 g ) which on flash chromatography and elution with $4 \%$ methanol in ethyl acetate gave the indoline $14(0.13 \mathrm{~g}, 68 \%)$ as a colourless solid which recrystallised from diethyl ether-chloroform; m.p. $122^{\circ} \mathrm{C}$; $\delta_{\mathrm{H}}(400 \mathrm{MHz}) 1.86(1 \mathrm{H}, \mathrm{m}, 1-\mathrm{H}), 2.31\left(3 \mathrm{H}, \mathrm{s}, 5-\mathrm{CH}_{3}\right), 2.44$ $(1 \mathrm{H}, \mathrm{m}, 1-\mathrm{H}), 3.40(1 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}), 3.78\left(3 \mathrm{H}, \mathrm{s}, \mathrm{OCH}_{3}\right), 3.8(1 \mathrm{H}$, $\mathrm{m}, 2-\mathrm{H}), 3.98(1 \mathrm{H}, \mathrm{q}, J 8,9 \mathrm{~b}-\mathrm{H}), 5.19(1 \mathrm{H}, \mathrm{s}, 4-\mathrm{H}), 5.48(1 \mathrm{H}, \mathrm{d}$, $J 8,9 \mathrm{c}-\mathrm{H}), 6.7(1 \mathrm{H}, \mathrm{dd}, J 8,3,7-\mathrm{H}), 6.8(1 \mathrm{H}, \mathrm{d}, J 3,9-\mathrm{H})$ and $6.9(1 \mathrm{H}, \mathrm{d}, J 8,6-\mathrm{H}) ; v_{\max }(\mathrm{KBr}) / \mathrm{cm}^{-1} 1640$ and $1610 ; \lambda_{\text {max }} / \mathrm{nm}$ $241\left(\varepsilon / \mathrm{dm}^{3} \mathrm{~mol}^{-1} \mathrm{~cm}^{-1} 11560\right)$ and 337 ( 14010 ); $m / z$ (EI) 256 $\left(\mathrm{M}^{+}, 100 \%\right.$ ) and 241 (35) (Found: C, 70.0; H, 6.2; N 10.7. $\mathrm{C}_{15} \mathrm{H}_{16} \mathrm{~N}_{2} \mathrm{O}_{2}$ requires $\mathrm{C}, 70.3 ; \mathrm{H}, 6.29 ; \mathrm{N}, 10.9 \%$ ).
( $\pm$ )-5-Benzyl 1-Ethyl 2-Acetylpentanedioate 17.-A solution of potassium tert-butoxide ( 0.012 mol ) in THF $\left(50 \mathrm{~cm}^{3}\right)$ was added over 30 min to a solution of ethyl acetoacetate $(6.5 \mathrm{~g}$, 0.05 mol ) and benzyl acrylate ${ }^{24}$ (from acrylyl chloride ${ }^{25}$ ) in THF ( $100 \mathrm{~cm}^{3}$ ) and heated at reflux for 3 h . Vigorous reaction
occurred on warming. The mixture was neutralised with acetic acid ( $15 \mathrm{~cm}^{3}$ ) and the tetrahydrofuran saturated with water ( $200 \mathrm{~cm}^{3}$ ). The aqueous solution was extracted with chloroform ( $3 \times 30 \mathrm{~cm}^{3}$ ) and the combined extracts washed with water and then dried $\left(\mathrm{MgSO}_{4}\right)$. The solvent was removed under reduced pressure to give a light brown oil which was distilled under reduced pressure yielding the benzyl ester 17 $(8.3 \mathrm{~g}, 57 \%)$ as a clear oil; b.p. $158-162^{\circ} \mathrm{C} / 0.1 \mathrm{mmHg} ; \delta_{\mathrm{H}} 1.25$ $\left(3 \mathrm{H}, \mathrm{t}, \mathrm{J} 7, \mathrm{CH}_{3} \mathrm{CH}_{2}\right), 2.15\left(2 \mathrm{H}, \mathrm{q}, \mathrm{J} 7, \mathrm{CHCH}_{2} \mathrm{CH}_{2}\right), 2.2$ ( $3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3} \mathrm{C}$ ), $2.4\left(2 \mathrm{H}, \mathrm{t}, J 7, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}=\mathrm{O}\right), 3.55(1 \mathrm{H}, \mathrm{t}, J 7$, $\mathrm{COCHCO}), 4.2\left(2 \mathrm{H}, \mathrm{q}, \mathrm{J} 7, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 5.1\left(2 \mathrm{H}, \mathrm{s}, \mathrm{OCH}_{2} \mathrm{Ph}\right)$ and $7.35(5 \mathrm{H}, \mathrm{s}, \mathrm{ArH}) ; v_{\text {max }}($ liq. film $) / \mathrm{cm}^{-1} 1727$ and $1709 ; m / z$ (FD) $292\left(\mathrm{M}^{+}, 100 \%\right)$ (Found: C, 65.8; H, 7.0. $\mathrm{C}_{16} \mathrm{H}_{20} \mathrm{O}_{5}$ requires $\mathrm{C}, 65.7 ; \mathrm{H}, 6.9 \%$ ).
( $\pm$ )-4-Ethoxycarbonyl-5-oxohexanoic Acid 18.-The above benzyl ester $17(10 \mathrm{~g}, 0.034 \mathrm{~mol})$ in absolute alcohol $\left(60 \mathrm{~cm}^{3}\right)$ was hydrogenated for 90 min at atmospheric pressure using $5 \%$ palladium on charcoal ( 0.6 g ). The reaction mixture was filtered and solvent removal from the filtrate under reduced pressure gave the acid 18 as a clear oil; $\delta_{\mathrm{H}} 1.3(3 \mathrm{H}, \mathrm{t}, J 7$, $\mathrm{CH}_{3} \mathrm{CH}_{2}$ ), $\left.2.18(2 \mathrm{H}, \mathrm{q}, J 7, \mathrm{CHCH})_{2}\right), 2.28(3 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{CH}_{3} \mathrm{CO}\right), 2.44\left(2 \mathrm{H}, \mathrm{t}, J 7, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CO}\right), 3.58(1 \mathrm{H}, \mathrm{t}, J 7$, $\mathrm{OCHCO}), 4.2\left(2 \mathrm{H}, \mathrm{q}, J 7, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right)$ and $12.84(1 \mathrm{H}, \mathrm{s}$, OH ) (Found: C, 53.3; H, 6.9. $\mathrm{C}_{19} \mathrm{H}_{14} \mathrm{O}_{5}$ requires C, $53.5 ; \mathrm{H}$, 7.0\%).
( $\pm$ )-4-Ethoxycarbonyl-N-[2-(1H-indol-3-yl)ethyl]-5-oxo-
hexanamide 19.-To a solution of the above acid $18(5 \mathrm{~g}, 0.025$ $\mathrm{mol})$ in THF $\left(75 \mathrm{~cm}^{3}\right)$ cooled to $0^{\circ} \mathrm{C}$, was added triethylamine $(2.53 \mathrm{~g}, 0.025 \mathrm{~mol})$ in THF ( $30 \mathrm{~cm}^{3}$ ) followed by ethyl chloroformate ( $2.71 \mathrm{~g}, 0.025 \mathrm{~mol}$ ) in THF ( $30 \mathrm{~cm}^{3}$ ). The solution was stirred at $0^{\circ} \mathrm{C}$ for 30 min and then tryptamine $(4 \mathrm{~g}, 0.025$ mol ) in THF ( $30 \mathrm{~cm}^{3}$ ) was added over 10 min , and the mixture warmed to $20^{\circ} \mathrm{C}$ and stirred for a further 16 h . The reaction was quenched with saturated aqueous sodium hydrogen carbonate and then extracted with dichloromethane $\left(3 \times 50 \mathrm{~cm}^{3}\right)$. The organic extracts were combined, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and then the solvent was removed under reduced pressure to give a brown oil which was purified by flash chromatography.

Elution with ethyl acetate gave the amide $19(6.7 \mathrm{~g}, 78 \%)$ as a light brown oil; $\delta_{\mathrm{H}} 1.24\left(3 \mathrm{H}, \mathrm{t}, J 7, \mathrm{CH}_{3} \mathrm{CH}_{2}\right), 2.13(4 \mathrm{H}, \mathrm{m}$, $\mathrm{CHCH}_{2} \mathrm{CH}_{2} \mathrm{CO}$ ), $2.19\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3} \mathrm{CO}\right), 2.99(2 \mathrm{H}, \mathrm{t}, J 7$, Ind$\left.\mathrm{CH}_{2}\right), 3.55(1 \mathrm{H}, \mathrm{t}, J 7, \mathrm{COCHCO}), 3.59(2 \mathrm{H}, \mathrm{q}, J 7$, Ind $\left.\mathrm{CH}_{2} \mathrm{CH}_{2}\right), 4.16\left(2 \mathrm{H}, \mathrm{q}, J 7, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 5.5(1 \mathrm{H}$, br s, NHCO), 7.05 ( $1 \mathrm{H}, \mathrm{s}$, Ind-2-H), 7.12 ( $1 \mathrm{H}, \mathrm{t}, J$ 8, Ind-5-H), 7.18 ( $1 \mathrm{H}, \mathrm{t}, J$, Ind-6-H), $7.32(1 \mathrm{H}, \mathrm{d}, J 8$, Ind-7-H), $7.60(1 \mathrm{H}, \mathrm{d}, J$ 8, Ind-4-H) and 8.15 ( $1 \mathrm{H}, \mathrm{br} \mathrm{s}$, Ind-NH); $v_{\text {max }}$ (liq. film) $/ \mathrm{cm}^{-1}$ 3375, 1730, 1705 and $1645 ; \lambda_{\text {max }} / \mathrm{nm} 228\left(\varepsilon / \mathrm{dm}^{3} \mathrm{~mol}^{-1} \mathrm{~cm}^{-1}\right.$ 22 350) 275 (7050) 282 (7400) and 291 (6370); $m / z$ (EI) 344 $\left(\mathrm{M}^{+}, 3 \%\right) 143$ (88) and 130 (80) (Found: $\mathrm{M}^{+} 344.1725$. $\mathrm{C}_{19} \mathrm{H}_{24} \mathrm{~N}_{2} \mathrm{O}_{4}$ requires $M, 344.1735$ ).
( $\pm$ )-4-Ethoxycarbonyl-N-[2-(5-methoxy-1H-indol-3-yl)ethyl ]-5-oxohexanamide 20.-To a solution of 4-ethoxycar-bonyl-5-oxohexanoic acid $(0.25 \mathrm{~g}, 1.24 \mathrm{mmol}$ ) in THF ( 20 $\mathrm{cm}^{3}$ ) at $0{ }^{\circ} \mathrm{C}$ under an atmosphere of nitrogen was added triethylamine ( $0.14 \mathrm{~g}, 1.36 \mathrm{mmol}$ ) in THF ( $5 \mathrm{~cm}^{3}$ ) followed by ethyl chloroformate ( $0.15 \mathrm{~g}, 1.36 \mathrm{mmol}$ ) in THF $\left(5 \mathrm{~cm}^{3}\right)$ and the solution was stirred for 15 min at $0^{\circ} \mathrm{C}$. 5-Methoxytryptamine ( $0.24 \mathrm{~g}, 1.24 \mathrm{mmol}$ ) in THF $\left(5 \mathrm{~cm}^{3}\right)$ was added over 5 $\min$ and the solution was warmed to $20^{\circ} \mathrm{C}$ and stirred for 24 h at which time filtration and solvent removal gave a brown oil $(0.65 \mathrm{~g})$. Chromatography and elution of this with ethyl acetate gave the amide 20 as a brown oil $(0.15 \mathrm{~g}, 32 \%)$; $\delta_{\mathrm{H}} 1.12(3 \mathrm{H}$, $\mathrm{t}, \mathrm{J} 7, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{CO}_{2}$ ), $2.15\left(4 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CONH}\right.$ ), 2.25 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3} \mathrm{CO}$ ), $2.95\left(2 \mathrm{H}, \mathrm{t}, J 7\right.$, Ind- $\mathrm{CH}_{2}$ ), $3.63(3 \mathrm{H}, \mathrm{m}$, Ind$\left.\mathrm{CH}_{2} \mathrm{CH}_{2}, \mathrm{COCHCO}\right), 3.88\left(3 \mathrm{H}, \mathrm{s}, \mathrm{OCH}_{3}\right), 4.17(2 \mathrm{H}, \mathrm{m}$,
$\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{CO}_{2}$ ), $5.62(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{NHCO}), 6.87(1 \mathrm{H}, \mathrm{dd}, J 8,2$, Ind-6-H), $7.02(2 \mathrm{H}, \mathrm{m}$, Ind-2-H and $4-\mathrm{H}), 7.26(1 \mathrm{H}, \mathrm{d}, J 8$, Ind-7-H) and 8.07 ( 1 H , br s, Ind-NH); $v_{\text {max }}$ (liq. film) $/ \mathrm{cm}^{-1}$ 1730, 1710 and 1650; $\lambda_{\text {max }} / \mathrm{nm} 224\left(\varepsilon / \mathrm{dm}^{3} \mathrm{~mol}^{-1} \mathrm{~cm}^{-1}\right.$ 24 100) 277 (6835) 292 (5400) and 310 (3810); $m / z$ (EI) 374 $\left(\mathrm{M}^{+}, 1 \%\right), 173$ (25), 160 (30) and 43 (100) (Found: C, 64.1; H, $6.9 ; \mathrm{N}, 7.35 . \mathrm{C}_{20} \mathrm{H}_{26} \mathrm{~N}_{2} \mathrm{O}_{5}$ requires C, $64.15 ; \mathrm{H}, 7.00 ; \mathrm{N}, 7.48 \%$ ).
( $1 \mathrm{R}^{*}, 12 \mathrm{bR} \mathrm{R}^{*}$ )- and ( $1 \mathrm{~S}^{*}, 12 \mathrm{bR} \mathrm{R}^{*}$ )-1-Ethoxycarbonyl-2,3,6,7-tetrahydro-12b-methyl-12H-indolo[2,3-a $]$ quinolizin-4-(1H)-one 21 and 22. -The amide $19(0.2 \mathrm{~g}, 0.58 \mathrm{mmol})$ in dry acetonitrile ( $20 \mathrm{~cm}^{3}$ ) was stirred at $20^{\circ} \mathrm{C}$ under an atmosphere of nitrogen. Trifluoracetic acid anhydride ( $0.12 \mathrm{~g}, 0.58 \mathrm{mmol}$ ) was added over 5 min and the mixture stirred for 48 h . The solvent was removed under reduced pressure to give a dark yellow oil which was purified by flash chromatography. Elution with chloroform gave the cis-indolo[2,3-a]quinolizinone $21(0.15 \mathrm{~g}$, $79 \%$ ) as colourless crystals (ethyl acetate); m.p. $237-238^{\circ} \mathrm{C}$; $\delta_{\mathrm{H}}$ $1.34\left(3 \mathrm{H}, \mathrm{t}, \mathrm{J} 7, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 1.8(3 \mathrm{H}, \mathrm{s}, 12 \mathrm{~b}-\mathrm{Me}), 2.2(2 \mathrm{H}, \mathrm{m}$, $\left.2-\mathrm{H}_{2}\right), 2.51(1 \mathrm{H}, \mathrm{m}, 3 \alpha-\mathrm{H}), 2.67(1 \mathrm{H}, \mathrm{dt}, J 18,4,3 \beta-\mathrm{H}), 2.71$ ( 1 H, ddd, $J 16,4,2,7 \beta-\mathrm{H}), 2.83(1 \mathrm{H}$, ddd, $J 16,10,5,7 \alpha-\mathrm{H})$, $3.01(2 \mathrm{H}, \mathrm{m}, 1-\mathrm{H}$ and $6 \beta-\mathrm{H}), 4.3\left(2 \mathrm{H}, \mathrm{q}, \mathrm{J}, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 5.07$ ( 1 H , ddd, $J 13,4,2,6 \alpha-\mathrm{H})$, $7.1(1 \mathrm{H}, \mathrm{t}, J 8$, Ind-5-H), $7.19(1 \mathrm{H}$, $\mathrm{t}, J 8$, Ind-6-H), $7.31(1 \mathrm{H}, \mathrm{d}, J 8$, Ind-7-H), $7.52(1 \mathrm{H}, \mathrm{d}, J 8$, Ind-$4-\mathrm{H}$ ) and 9.2 ( $1 \mathrm{H}, \mathrm{br} \mathrm{s}$, Ind-N-H); $\delta_{\mathrm{C}}$ 14.0, 20.3, 21.1, 22.4, 30.2, 38.3, 49.9, 59.7, 62.1, 109.1, 111.3, 118.5, 119.8, 122.6, 125.9, 136 ( 2 signals), 171.4 and $174 ; v_{\text {max }}\left(\mathrm{CHCl}_{3}\right) / \mathrm{cm}^{-1} 3430,1720$ and 1633; $\lambda_{\text {max }} / \mathrm{nm} 223\left(\varepsilon / \mathrm{dm}^{3} \mathrm{~mol}^{-1} \mathrm{~cm}^{-1} 37600\right) 273$ (8200) 280 (8140) and 290.5 (6300); $m / z(E I) 326\left(\mathrm{M}^{+}, 60 \%\right), 311$ (100) and 183 (45) (Found: $\mathrm{M}^{+}$326.1636. $\mathrm{C}_{19} \mathrm{H}_{22} \mathrm{~N}_{2} \mathrm{O}_{3}$ requires $M$, 326.1636).

Further elution gave the trans-indolo[2,3-a]quinolizinone 22 as colourless crystals (ethyl acetate) ( $30 \mathrm{mg}, 16 \%$ ); m.p. $242-$ $243{ }^{\circ} \mathrm{C} ; \delta_{\mathrm{H}} 0.72\left(3 \mathrm{H}, \mathrm{t}, J 7, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 1.76(3 \mathrm{H}, \mathrm{s}, 12 \mathrm{~b}-\mathrm{Me})$, 2.12 and $2.34\left(2 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}_{2}\right), 2.58$ ( 1 H, ddd, $J 16,7,2,3 \alpha-\mathrm{H}$ ), $2.76\left(3 \mathrm{H}, \mathrm{m}, 3 \beta-\mathrm{H}\right.$ and $\left.7-\mathrm{CH}_{2}\right), 3.0(1 \mathrm{H}, \mathrm{ddd}, J 12,9,5,6 \beta-\mathrm{H})$, $3.22(1 \mathrm{H}, \mathrm{t}, J 4,1, \beta-\mathrm{H}), 3.77\left(2 \mathrm{H}, \mathrm{q}, \mathrm{J} 7, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 5.16$ $(1 \mathrm{H}, \mathrm{dt}, J 12,4,6 \alpha-\mathrm{H}), 7.12(1 \mathrm{H}, \mathrm{t}, J 8$, Ind-5-H), $7.18(1 \mathrm{H}, \mathrm{t}$, $J 8$, Ind-6-H), $7.32(1 \mathrm{H}, \mathrm{d}, J 8$, Ind-7-H), 7.48 ( $1 \mathrm{H}, \mathrm{d}, J 8$, Ind-$4-\mathrm{H})$ and $8.2\left(1 \mathrm{H}\right.$, br s, Ind-NH); $\delta_{\mathrm{C}} 13.5,19.2,20.8,22.1,28.1$, 37.2,48,58.4,60.7, 110.0, 111.0, 118.4, 119.6, 122.0, 126.3, 135.0, 137.0, 169.7 and 171.4; $v_{\max }\left(\mathrm{CHCl}_{3}\right) / \mathrm{cm}^{-1} 3450,1725$ and 1625 ; $\lambda_{\text {max }} / \mathrm{nm} 224.5\left(\varepsilon / \mathrm{dm}^{3} \mathrm{~mol}^{-1} \mathrm{~cm}^{-1} 35000\right) \quad 275$ ( 6800 ), 282 ( 6800 ) and 291 ( 5700 ); $m / z(E I) 326\left(\mathrm{M}^{+}, 30 \%\right) 311$ (100) and 183 (30); (FD) 326 (100) (Found: $\mathrm{M}^{+} 326.1633 . \mathrm{C}_{19} \mathrm{H}_{22} \mathrm{~N}_{2} \mathrm{O}_{3}$ requires $M, 326.1636$ ).

Reaction of the Amide 19 with Phosphoryl Chloride.-The amide $19(0.125 \mathrm{~g}, 0.29 \mathrm{mmol})$ was stirred in dry acetonitrile ( $10 \mathrm{~cm}^{3}$ ) at $20^{\circ} \mathrm{C}$ under an atmosphere of nitrogen. Phosphoryl chloride ( $31 \mathrm{~mm}^{3}, 0.29 \mathrm{mmol}$ ) was added and the mixture was stirred for 30 h . The reaction was quenched with dilute aqueous ammonia and the mixture extracted with dichloromethane ( $3 \times 20 \mathrm{~cm}^{3}$ ). The combined extracts were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and the solvent was removed under reduced pressure to give a pale yellow solid. Crystallisation from ethyl acetate gave the cis-indoloquinolizinone 21 ( $94 \mathrm{mg}, 99 \%$ ). Analysis of the crude material by ${ }^{1} \mathrm{H}$ NMR spectroscopy showed a trace amount of the trans-isomer 22.
(1R*,12bR*)-1-Ethoxycarbonyl-2,3,6,7-tetrahydro-9-meth-oxy-12b-methyl-12H-indolo[2,3-a]quinolizin- $4(1 \mathrm{H})$-one 23 .Phosphoryl chloride ( $31 \mathrm{~mm}^{3}, 0.27 \mathrm{mmol}$ ) was added over 5 min , to a solution of the 5 -methoxyamide $20(0.1 \mathrm{~g}, 0.27 \mathrm{mmol})$ in dry acetonitrile ( $10 \mathrm{~cm}^{3}$ ) under an atmosphere of nitrogen at $20^{\circ} \mathrm{C}$. After being stirred for 20 h , the reaction was quenched with saturated aqueous sodium hydrogen carbonate $\left(10 \mathrm{~cm}^{3}\right)$ and the mixture extracted with dichloromethane ( $3 \times 10 \mathrm{~cm}^{3}$ ).

The extracts were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and the solvent was removed under reduced pressure to give a green oil ( 78 mg ) which was purified by flash chromatography. Elution with $99 \%$ dichloromethane in methanol gave the title compound 23 as a clear oil ( $63 \%$ ) which crystallised from diethyl ether; m.p. $161^{\circ} \mathrm{C} ; \delta_{\mathrm{H}} 1.32\left(3 \mathrm{H}, \mathrm{t}, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{CO}_{2}\right), 1.78(3 \mathrm{H}, \mathrm{s}, 12 \mathrm{~b}-\mathrm{Me})$, $2.19\left(2 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}_{2}\right), 2.51(1 \mathrm{H}, \mathrm{m}, 3 \alpha-\mathrm{H}), 2.67(1 \mathrm{H}, \mathrm{dt}, J 18,4$, $3 \beta-\mathrm{H}), 2.70(1 \mathrm{H}$, ddd, $J 16,4,2,7 \beta-\mathrm{H}), 2.80(1 \mathrm{H}$, ddd, $J 16,10$, $5,7 \alpha-\mathrm{H}), 3.0(2 \mathrm{H}, \mathrm{m}, 1 \alpha-\mathrm{H}$ and $6 \beta-\mathrm{H}), 3.84(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}), 4.29$ $\left(2 \mathrm{H}, \mathrm{q}, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{CO}_{2}\right), 5.07(1 \mathrm{H}$, ddd, $J 12,5,2,6 \alpha-\mathrm{H}) 6.84$ ( 1 H, dd, $J 8,2$, Ind-6-H), 6.93 ( $1 \mathrm{H}, \mathrm{d}, J 2$, Ind-4-H), 7.22 ( $1 \mathrm{H}, \mathrm{d}, J \mathrm{~B}$, Ind-7-H) and $9.06(1 \mathrm{H}$, br s, Ind-NH); $v_{\max }\left(\mathrm{CHCl}_{3}\right) / \mathrm{cm}^{-1} 1715$ and $1625 ; \lambda_{\max } / \mathrm{nm} 226\left(\varepsilon / \mathrm{dm}^{3} \mathrm{~mol}^{-1}\right.$ $\mathrm{cm}^{-1} 33590$ ), 276 (9415), 296 (6110) and 309 (4070); $m / z$ (EI) $356\left(\mathrm{M}^{+}, 40 \%\right) 341(100)$ and 213 (45) (Found: C, 67.6; H, 6.8; $\mathrm{N}, 8.1 . \mathrm{C}_{20} \mathrm{H}_{24} \mathrm{~N}_{2} \mathrm{O}_{4}$ requires $\mathrm{C}, 67.4 ; \mathrm{H}, 6.8 ; \mathrm{N}, 7.9 \%$ ).

Cyclisation of the Amide 19 with Excess of Trifluoracetic Acid Anhydride.-To a solution of the amide 19 ( $45 \mathrm{mg}, 0.133$ $\mathrm{mmol})$ in dry benzene $\left(10 \mathrm{~cm}^{3}\right)$ at $20^{\circ} \mathrm{C}$ under an atmosphere of nitrogen was added trifluoracetic acid anhydride (TFAA) $\left(0.2 \mathrm{~cm}^{3}, 1.33 \mathrm{mmol}\right.$ ) over 5 min . An orange-brown solution formed immediately and the solution was stirred at $20^{\circ} \mathrm{C}$ for 72 h . Removal of solvent under reduced pressure gave a redbrown oil ( 70 mg ) which was purified by flash chromatography. Elution with ethyl acetate-methanol (1:1) gave the ( $\pm$ )-pyrrolo[2,3-d]carbazole 27 as a yellow solid ( $28 \mathrm{mg}, 50 \%$ ); $\delta_{\mathrm{H}}$ $1.42\left(3 \mathrm{H}, \mathrm{t}, \mathrm{J} 7, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{CO}_{2}\right), 2.5$ and $2.84\left(2 \mathrm{H}, \mathrm{m}, 1-\mathrm{H}_{2}\right), 3.1$ $\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3} \mathrm{CO}\right), 4.46\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{CO}_{2}\right), 4.86$ and $5.44(2$ $\left.\mathrm{H}, \mathrm{m}, 2-\mathrm{H}_{2}\right), 6.32$ and $6.44(0.3 \mathrm{H}$ and $0.7 \mathrm{H}, 2 \times \mathrm{s}, 6 \mathrm{a}-\mathrm{H}), 7.02$ $(1 \mathrm{H}, \mathrm{d}, J 8,11-\mathrm{H}), 7.23(1 \mathrm{H}, \mathrm{t}, J 8,10-\mathrm{H}), 7.48(1 \mathrm{H}, \mathrm{t}, J 8$, $9-\mathrm{H}), 7.68(1 \mathrm{H}, \mathrm{d}, J 7.5,4-\mathrm{H}), 8.25(1 \mathrm{H}, \mathrm{d}, J 8,8-\mathrm{H})$ and 8.87 $(1 \mathrm{H}, \mathrm{d}, J 7.5,5-\mathrm{H}) ; m / z(\mathrm{FD}) 420\left(\mathrm{M}^{+}, 100 \%\right) 392(85)$ and 324 (50). Further data on this product was not obtained due to its instability.
( $\pm$ )-N-[2-(1-Methylindol-3-yl)ethyl]-3-(1,3-dioxobutyl)-piperidin-2-one 29.--To a stirred solution of diisopropylamine ( $0.55 \mathrm{~cm}^{3}, 3.9 \mathrm{mmol}$ ) in THF ( $4 \mathrm{~cm}^{3}$ ) under an atmosphere of nitrogen at $0-4^{\circ} \mathrm{C}$ was added a solution of butyllithium in hexane ( $0.822 \mathrm{~mol} \mathrm{dm}^{-3} ; 4.74 \mathrm{~cm}^{3}, 3.9 \mathrm{mmol}$ ). After 15 min , at $0-4^{\circ} \mathrm{C}$ the solution was cooled to $-78^{\circ} \mathrm{C}$. $N$-[2-(1-Methyl-indol-3-yl)ethyl]piperidin-2-one ${ }^{26}(0.5 \mathrm{~g}, 1.95 \mathrm{mmol})$ in THF $\left(5 \mathrm{~cm}^{3}\right)$ was added dropwise and the mixture stirred at $-78^{\circ} \mathrm{C}$ for 30 min . Then diketene ( $153 \mathrm{~mm}^{3}, 1.9 \mathrm{mmol}$ ) in THF (4 $\mathrm{cm}^{3}$ ) was added over 5 min and stirred for a further 2 h at $-78^{\circ} \mathrm{C}$. The mixture was then warmed to $0^{\circ} \mathrm{C}$ and the reaction quenched with $10 \%$ hydrochloric acid to pH 2 . The mixture was extracted with dichloromethane ( $3 \times 30 \mathrm{~cm}^{3}$ ) and the combined extracts were washed with water and dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and then the solvent was removed under reduced pressure to give a brown oil ( 0.8 g ). Flash chromatography, eluting with $2 \%$ methanol in dichloromethane gave the dione 29 as a clear oil, shown to be a mixture of predominantly two enol forms $(0.23 \mathrm{~g}, 35 \%) ; \delta_{\mathrm{H}} 1.70\left(2 \mathrm{H}, \mathrm{m}, 5-\mathrm{H}_{2}\right), 1.90$ and $2.15(1 \mathrm{H}$, $2 \times \mathrm{m}, 4-\mathrm{H}$ enol c), $2.08\left(1.5 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3} \mathrm{CO}\right.$, enol c), $2.28(1.5 \mathrm{H}$, $\mathrm{s}, \mathrm{CH}_{3} \mathrm{CO}$, enol b), 2.32 ( $1 \mathrm{H}, \mathrm{t}, 4-\mathrm{H}$ enol b), $3.06(2 \mathrm{H}, \mathrm{t}, J 7$, Ind- $\mathrm{CH}_{2}$ ), $3.20\left(2 \mathrm{H}, \mathrm{t}, J 4,6-\mathrm{H}_{2}\right), 3.34(0.5 \mathrm{H}, \mathrm{t}, J 7,3-\mathrm{H}$ enol c), $3.36\left(1 \mathrm{H}, \mathrm{s}, \mathrm{COCH}_{2} \mathrm{COCH}_{3}\right.$, enol b) $3.68(2 \mathrm{H}, \mathrm{t}, J 7$, Ind$\left.\mathrm{CH}_{2} \mathrm{CH}_{2}\right), 3.75\left(3 \mathrm{H}, \mathrm{s}, \mathrm{NCH}_{3}\right), 5.72(0.5 \mathrm{H}, \mathrm{s}, \mathrm{COCH}=\mathrm{COH}$ enol c), $6.93(1 \mathrm{H}, 2 \times \mathrm{s}$, Ind-2-H), $7.12(1 \mathrm{H}, \mathrm{t}, J 8$, Ind-5-H), $7.24(1 \mathrm{H}, \mathrm{t}, J 8$, Ind-6-H), 7.31 ( $1 \mathrm{H}, \mathrm{d}, J 8$, Ind-7-H), $7.66(1 \mathrm{H}$, m , Ind-4-H), $15.20(0.5 \mathrm{H}$, s, enol-OH b) and $15.36(0.5 \mathrm{H}$, br s, enol-OH c); $v_{\max }($ film $) / \mathrm{cm}^{-1} 1720,1635$ and $1600 ; \lambda_{\max } / \mathrm{nm} 226$ $\left(\varepsilon / \mathrm{dm}^{3} \mathrm{~mol}^{-1} \mathrm{~cm}^{-1} 35520\right)$ and 275 ( 13640 ); $m / z$ (EI) $340\left(\mathrm{M}^{+}\right.$, $80 \%$ ), 298 (10), 256 (25), 157 (100) and 144 (60) (Found: C, $70.7 ; \mathrm{H}, 7.4 ; \mathrm{N}, 8.0 . \mathrm{C}_{20} \mathrm{H}_{24} \mathrm{~N}_{2} \mathrm{O}_{3}$ requires $\mathrm{C}, 70.56 ; \mathrm{H}, 7.11$; N, $8.23 \%$ ).
( $\pm$ )-3-Benzyloxycarbonyl-N-[2-(1-methylindol-3-yl)ethyl]-piperidin-2-one 32.-To a solution of diisopropylamine (1.2 $\mathrm{cm}^{3}, 8.6 \mathrm{mmol}$ ) in THF ( $5 \mathrm{~cm}^{3}$ ) under an atmosphere of nitrogen at $0-4^{\circ} \mathrm{C}$ was added a solution of butyllithium in hexane ( $0.822 \mathrm{~mol} \mathrm{~cm}^{-3} ; 10.5 \mathrm{~cm}^{3}, 8.6 \mathrm{mmol}$ ). The solution was stirred at $0-4^{\circ} \mathrm{C}$ for 20 min and then cooled to $-78^{\circ} \mathrm{C}$. The lactam 31 ( $2.0 \mathrm{~g}, 7.8 \mathrm{mmol}$ ) in THF ( $15 \mathrm{~cm}^{3}$ ) and hexamethylphosphoramide $\left(4 \mathrm{~cm}^{3}\right)$ was added dropwise, with stirring and then the solution stirred for 30 min at $-78^{\circ} \mathrm{C}$. Benzyl chloroformate ( $1.44 \mathrm{~cm}^{3}, 8.6 \mathrm{mmol}$ ) was added rapidly to the brown solution at $-78^{\circ} \mathrm{C}$ and a clear solution formed after 5 min which was stirred for 2 h at $-78^{\circ} \mathrm{C}$ and then slowly warmed to $0^{\circ} \mathrm{C}$. The reaction was quenched with dilute hydrochloric acid and extracted with dichloromethane $(3 \times 30$ $\mathrm{cm}^{3}$ ). The combined organic extracts were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and the solvent was removed under reduced pressure to give a clear oil which on flash chromatography and elution with dichloro-methane-ethyl acetate ( $95: 5$ ) gave the benzyl ester 32 as a clear oil $(2.23 \mathrm{~g}, 73 \%) ; \delta_{\mathrm{H}} 1.64$ and $1.82\left(2 \mathrm{H}, \mathrm{m}, 5-\mathrm{H}_{2}\right), 2.02$ ( $2 \mathrm{H}, \mathrm{m}, 4-\mathrm{H}_{2}$ ) $3.03\left(2 \mathrm{H}, \mathrm{t}, J 7\right.$, Ind- $\left.\mathrm{CH}_{2}\right), 3.14\left(2 \mathrm{H}, \mathrm{m}, 6-\mathrm{H}_{2}\right)$, $3.48(1 \mathrm{H}, \mathrm{t}, J 7,3-\mathrm{H}), 3.62$ and $3.68(2 \mathrm{H}, 2 \times \mathrm{t}, J 7$, Ind- $\left.\mathrm{CH}_{2} \mathrm{CH}_{2}\right), 3.70(3 \mathrm{H}, \mathrm{s}, \mathrm{NMe}), 5.22\left(2 \mathrm{H}, \mathrm{s}, \mathrm{COCH}_{2} \mathrm{Ph}\right)$, $6.68(1 \mathrm{H}, \mathrm{s}$, Ind-2-H), $7.10(1 \mathrm{H}, \mathrm{t}, J 8$, Ind-5-H), $7.22(1 \mathrm{H}, \mathrm{t}$, $J 8$, Ind-6-H), $7.30-7.50(6 \mathrm{H}, \mathrm{m}, \mathrm{Ar}$ and Ind-7-H) and 7.65 ( $1 \mathrm{H}, \mathrm{d}, J 8$, Ind-4-H); $v_{\max }($ film $) / \mathrm{cm}^{-1} 1735$ and $1640 ; \lambda_{\text {max }} / \mathrm{nm}$ $226\left(\varepsilon / \mathrm{dm}^{3} \mathrm{~mol}^{-1} \mathrm{~cm}^{-1} 36560\right.$ ), 280 (5625) and 289 (6100); $m / z$ (FD) 390 ( $100 \%$ ) (Found: C, 73.7; H, 7.0; N, 7.4. $\mathrm{C}_{24} \mathrm{H}_{26} \mathrm{~N}_{2} \mathrm{O}_{3}$ requires $\mathrm{C}, 73.82 ; \mathrm{H}, 6.7 ; \mathrm{N}, 7.18 \%)$.
( $\pm$ )-3-Benzyloxycarbonyl-N-[2-(1-methylindol-3-yl)ethyl]-3-(1-oxo-2-methoxycarbonylethyl)piperidin-2-one 33.-To a suspension of sodium hydride ( $28 \mathrm{mg}, 1.16 \mathrm{mmol}$ ) (washed 3 times with benzene) in benzene ( $5 \mathrm{~cm}^{3}$ ) at $20^{\circ} \mathrm{C}$ under an atmosphere of nitrogen was added the benzyl ester $32(0.3 \mathrm{~g}, 0.77 \mathrm{mmol})$ in benzene $\left(5 \mathrm{~cm}^{3}\right)$ over 5 min . The solution was stirred for 30 $\min$ at $20^{\circ} \mathrm{C}$, then methoxycarbonylacetyl chloride (freshly prepared) was added dropwise and the solution was stirred at $20^{\circ} \mathrm{C}$ for 1 h . TLC showed that no reaction had occurred and the solution was gently heated at reflux for 7 h . The mixture was then cooled to $0^{\circ} \mathrm{C}$, acidified with dilute hydrochloric acid and extracted with dichloromethane $\left(3 \times 20 \mathrm{~cm}^{3}\right)$. The combined extracts were washed with water and dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and then the solvent was removed under reduced pressure to give a brown oil. Flash chromatography and elution with ethyl acetate-dichloromethane (2:98) gave the required ester 33 as a clear oil $(0.18 \mathrm{~g}, 48 \%) ; \delta_{\mathrm{H}} 1.7(2 \mathrm{H}, \mathrm{m}, 5-\mathrm{H}), 2.08$ and $2.60(2 \mathrm{H}$, $\left.\mathrm{m}, 4-\mathrm{H}_{2}\right), 3.02\left(2 \mathrm{H}, \mathrm{m}\right.$, Ind- $\left.\mathrm{CH}_{2}\right), 3.16(2 \mathrm{H}, \mathrm{m}, 6-\mathrm{H}), 3.64$ ( $2-\mathrm{H}, \mathrm{m}$, Ind- $\mathrm{CH}_{2} \mathrm{CH}_{2}$ ), 3.66 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}$ ), 3.71 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{NMe}$ ), 3.72 and $4.14\left(2 \mathrm{H}, \mathrm{ABq}, \mathrm{J} 16, \mathrm{COCH}_{2} \mathrm{CO}_{2} \mathrm{Me}\right), 5.25(2 \mathrm{H}$, $\left.\mathrm{ABq}, J 13, \mathrm{OCH}_{2} \mathrm{Ph}\right), 6.86(1 \mathrm{H}, \mathrm{s}$, Ind-2-H), $7.11(1 \mathrm{H}, \mathrm{t}, J 8$, Ind-5-H), $7.22(1 \mathrm{H}, \mathrm{t}, J 8$, Ind-6-H), $7.3(1 \mathrm{H}, \mathrm{d}, J 8$, Ind-7-H), $7.36(5 \mathrm{H}, \mathrm{m}, \mathrm{ArH})$ and $7.63(1 \mathrm{H}, \mathrm{d}, J \mathrm{8}$, Ind-4-H); $v_{\text {max }}($ film $) / \mathrm{cm}^{-1} 1770-1710$ and $1640 ; \lambda_{\text {max }} / \mathrm{nm} 226\left(\varepsilon / \mathrm{dm}^{3} \mathrm{~mol}^{-1}\right.$ $\mathrm{cm}^{-1} 35530$ ) and 289 (5520); $m / z$ (EI) 490 ( $\mathrm{M}^{+}, 35 \%$ ), 459 (5), 390 (35), 157 (98) and 144 (100) (Found: C, 68.6; H, 6.5; N, 5.4. $\mathrm{C}_{28} \mathrm{H}_{30} \mathrm{~N}_{2} \mathrm{O}_{6}$ requires C, $68.6 ; \mathrm{H}, 6.2 ; \mathrm{N}, 5.7 \%$ ).
( $\pm$ )- N -[2-(1-Methylindol-3-yl)ethyl]-3-(2-methoxycarbonyl-1-oxoethyl)piperidin-2-one 30.--To the diester $33(0.25 \mathrm{~g}, 0.51$ mmol), in methanol $\left(10 \mathrm{~cm}^{3}\right)$ was added ammonium formate $(0.13 \mathrm{~g}, 2.04 \mathrm{mmol})$ and the mixture hydrogenated for 90 min over $10 \%$ palladium on charcoal ( 50 mg ). The catalyst was filtered off and water ( $50 \mathrm{~cm}^{3}$ ) added to the filtrate. The mixture was then extracted with dichloromethane $(3 \times 30$ $\left.\mathrm{cm}^{3}\right)$ and the combined extracts were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and then the solvent was removed under reduced pressure to give a light brown oil. Flash chromatography, elution with ethyl acetatehexane ( $1: 1$ ), gave the title ester 30 as a clear oil ( $0.116 \mathrm{~g}, 64 \%$ )
in a mixture of keto and enol forms; $\delta_{\mathrm{H}} 1.8\left(2 \mathrm{H}, \mathrm{m}, 5-\mathrm{H}_{2}\right), 2.18$ ( $1 \mathrm{H}, \mathrm{m}, 4-\mathrm{H}$, keto form), 2.36 ( $1 \mathrm{H}, \mathrm{t}, J 6,4-\mathrm{H}$, enol form), 3.04 ( $2 \mathrm{H}, \mathrm{m}$, Ind- $\mathrm{CH}_{2}$ ), $3.21(2 \mathrm{H}, \mathrm{t}, J 6,6-\mathrm{H}), 3.31(1 \mathrm{H}, \mathrm{s}$, $\mathrm{COCH}_{2} \mathrm{CO}$, enol form), $3.56(0.5 \mathrm{H}, \mathrm{t}, J 7,3-\mathrm{H}$, keto form), $3.67\left(2 \mathrm{H}, \mathrm{t}, \mathrm{J} 7\right.$, Ind- $\left.\mathrm{CH}_{2} \mathrm{CH}_{2}\right), 3.74\left(6 \mathrm{H}, 2 \times \mathrm{s}, \mathrm{NCH}_{3}\right.$, $\left.\mathrm{OCH}_{3}\right), 3.85\left(1 \mathrm{H}, \mathrm{ABq}, J 15, \mathrm{COCH}_{2} \mathrm{CO}\right.$ keto form), 6.89 and $6.92(1 \mathrm{H}, 2 \times \mathrm{s}$, Ind-2-H), $7.12(1 \mathrm{H}, \mathrm{t}, J 8$, Ind-5-H), $7.23(1 \mathrm{H}$, $\mathrm{t}, J 8$, Ind-6-H), $7.30(1 \mathrm{H}, \mathrm{d}, J \mathrm{8}$, Ind-7-H), 7.62 and 7.64 ( $1 \mathrm{H}, 2 \times \mathrm{d}, J$, Ind-4-H) and $15.2(0.5 \mathrm{H} \mathrm{s}$, enol-OH); $v_{\text {max }}$ (liquid) $/ \mathrm{cm}^{-1} 1745$ and $1635 ; \lambda_{\text {max }} / \mathrm{nm} 225\left(\varepsilon / \mathrm{dm}^{3} \mathrm{~mol}^{-1}\right.$ $\mathrm{cm}^{-1} 39100$ ) and 269 ( 13920 ); $m / z$ (EI) 356 ( $\mathrm{M}^{+}, 40 \%$ ), 325 (30), 283 (11), 157 (100) and 144 (92) (Found: $\mathbf{M}^{+} 356.1741$. $\mathrm{C}_{20} \mathrm{H}_{24} \mathrm{~N}_{2} \mathrm{O}_{4}$ requires $M, 356.1736$ ).
( $2 \mathrm{~S}^{*}, 3 \mathrm{R}^{*}, 12 \mathrm{R}^{*}$ )-3-Acetyl-5-deethyl-5,19-didehydro-1-methyl4 -oxoaspidospermidine 34 .--The dione 29 ( $50 \mathrm{mg}, 0.15 \mathrm{mmol}$ ) in dichloromethane $\left(5 \mathrm{~cm}^{3}\right)$ at $20^{\circ} \mathrm{C}$, under an atmosphere of nitrogen was treated dropwise with trifluoroacetic acid anhydride (TFAA) $\left(0.215 \mathrm{~cm}^{3}, 1.5 \mathrm{mmol}\right)$. After being stirred at $20^{\circ} \mathrm{C}$ for 24 h TLC analysis showed no starting material and a major product. The mixture was added to dilute aqueous sodium hydrogen carbonate and then extracted with diethyl ether ( $3 \times 20 \mathrm{~cm}^{3}$ ). The combined diethyl ether extracts were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and then the solvent was removed under reduced pressure to give a yellow oil. Flash chromatography, eluting with ethyl acetate gave the title didehydroaspidospermidine 34 as yellow crystals ( $28 \mathrm{mg}, 60 \%$ ); m.p. ${ }^{158-166}{ }^{\circ} \mathrm{C}$ (diethyl ether-hexane); $\delta_{\mathrm{H}} 1.8-2.4\left(6 \mathrm{H}, \mathrm{m}, 6-, 7-\right.$ and $\left.11-\mathrm{H}_{2}\right)$, $2.22\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3} \mathrm{CO}\right), 2.64\left(3 \mathrm{H}, \mathrm{s}, \mathrm{NCH}_{3}\right), 3.30(1 \mathrm{H}$, ddd, $J 12,8,4,8-\mathrm{H}$ or $10-\mathrm{H}$ ), $3.4(2 \mathrm{H}, \mathrm{m}$ and d, $J 8,3-\mathrm{H}$ and $8-\mathrm{H}$ or $10-\mathrm{H}), 3.5(1 \mathrm{H}, \mathrm{d}, \mathrm{t}, J 13,4,8-\mathrm{H}$ or $10-\mathrm{H}), 3.74$ ( 1 H, ddd, $J 12,10,6,8-\mathrm{H}$ or $10-\mathrm{H}$ ), $4.44(1 \mathrm{H}, \mathrm{d}, J 8,2-\mathrm{H}), 6.5$ $(1 \mathrm{H}, \mathrm{d}, J 8,17-\mathrm{H}), 6.72(1 \mathrm{H}, \mathrm{t}, J 8,15-\mathrm{H}), 6.88(1 \mathrm{H}, \mathrm{d}, J 8$, $14-\mathrm{H})$ and $7.18(1 \mathrm{H}, \mathrm{t}, J 8,16-\mathrm{H})$; $\delta_{\mathrm{C}} 17.9,21.5,32.3,32.8,37.6$, $45.0,50.6,52.3,58.2,70.8,100.9,107.9,118.2,121.1,128.9$, $132.2,148.8,163.2,186.5$ and $208.7 ; v_{\max }\left(\mathrm{CHCl}_{3}\right) / \mathrm{cm}^{-1} 1710$, 1645 and $1570 ; \lambda_{\text {max }} / \mathrm{nm} 253\left(\varepsilon / \mathrm{dm}^{3} \mathrm{~mol}^{-1} \mathrm{~cm}^{-1} 14030\right)$ and 308 (23580); $m / z$ (EI) 322 (M ${ }^{+}, 100 \%$ ), 307 (25), 279 (95), 157 (56) and 144 (60) (FD) 322 (M ${ }^{+}$, 100) (Found: C, 74.1; H, 7.1; N, 8.9. $\mathrm{C}_{20} \mathrm{H}_{22} \mathrm{~N}_{2} \mathrm{O}_{2}$ requires $\mathrm{C}, 74.5 ; \mathrm{H}, 6.9 ; \mathrm{N}, 8.7 \%$ ).
(2S*,5R*,12R*,19S*)-3-Acetyl-5-deethyl-1-methyl-4-oxoaspidospermidine 35.--The dehydroaspidospermidine 34 (42 $\mathrm{mg}, 0.13 \mathrm{mmol})$ in methanol $\left(5 \mathrm{~cm}^{3}\right)$ at $20^{\circ} \mathrm{C}$ under nitrogen was treated with trifluoroacetic acid $\left(0.2 \mathrm{~cm}^{3}\right)$ and sodium cyanoborohydride ( $29 \mathrm{mg}, 0.46 \mathrm{~mol}$ ) and the mixture stirred for 5 min at $20^{\circ} \mathrm{C}$. The reaction was then quenched with $10 \%$ aqueous sodium hydrogen carbonate, and the mixture extracted with dichloromethane ( $3 \times 10 \mathrm{~cm}^{3}$ ). The combined extracts were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and the solvent was removed under reduced pressure to give a light brown oil. Flash chromatography and elution with chloroform gave the title deethylaspidospermidine 35 as a colourless solid ( $7 \mathrm{mg}, 17 \%$ ); m.p. $57-58{ }^{\circ} \mathrm{C}$; $\delta_{\mathrm{H}} 1.4-2.0\left(5 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2}\right.$ and CH$), 2.14(3 \mathrm{H}, \mathrm{s}$, $\mathrm{CH}_{3} \mathrm{CO}$ ), $2.22(1 \mathrm{H}, \mathrm{d}, J 3,19-\mathrm{H}), 2.36\left(3 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2}\right.$ and CH$)$, $2.54\left(3 \mathrm{H}, \mathrm{s}, \mathrm{NCH}_{3}\right), 2.91(1 \mathrm{H}, \mathrm{br} \mathrm{s}, 8-\mathrm{H}$ or $10-\mathrm{H}), 3.08(2 \mathrm{H}, \mathrm{m}$, $\mathrm{CH}_{2}$ ), $3.88(1 \mathrm{H}, \mathrm{s}, 2-\mathrm{H}), 6.48(1 \mathrm{H}, \mathrm{d}, J 8,17-\mathrm{H}), 6.80(1 \mathrm{H}$, $\mathrm{t}, J 8,15-\mathrm{H}), 7.14(2 \mathrm{H}, \mathrm{m}, 14-\mathrm{H}$ and $16-\mathrm{H})$ and $16.27(1 \mathrm{H}, \mathrm{s}$, enol); $v_{\text {max }}\left(\mathrm{CHCl}_{3}\right) / \mathrm{cm}^{-1} 1730 ; \lambda_{\text {max }} / \mathrm{nm} 256\left(\varepsilon / \mathrm{dm}^{3} \mathrm{~mol}^{-1} \mathrm{~cm}^{-1}\right.$ 16 200) and 284 ( 11771 ); m/z (EI) 324 ( $\mathrm{M}^{+}, 70 \%$ ), 296 (30) and 279 (43) (Found: $\mathrm{M}^{+} 324.1818 . \mathrm{C}_{20} \mathrm{H}_{24} \mathrm{~N}_{2} \mathrm{O}_{2}$ requires M, 324.1838).
Elution with $2.5 \%$ methanol in chloroform gave the less polar alcohol, ( $\pm$ ) -36 or $37(8.8 \mathrm{mg}, 21 \%)$ as a slightly impure colourless solid; m.p. $74-77^{\circ} \mathrm{C}$; $\delta_{\mathrm{H}} 1.20(3 \mathrm{H}, \mathrm{d}, J 7$, $\left.\mathrm{CH}_{3} \mathrm{CHOH}\right), 1.9-2.6\left(6 \mathrm{H}, \mathrm{m}, 3 \times \mathrm{CH}_{2}\right), 2.48(1 \mathrm{H}, \mathrm{dd}, J 6.5$, $3,3-\mathrm{H}), 2.93\left(3 \mathrm{H}, \mathrm{s}, \mathrm{NCH}_{3}\right), 3.3-3.8\left(4 \mathrm{H}, \mathrm{m}, 8-\mathrm{H}_{2}\right.$ and $10-\mathrm{H}_{2}$ ), $3.45(1 \mathrm{H}, \mathrm{d}, J 6.5,2-\mathrm{H}), 4.07\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{3} \mathrm{CHOH}\right), 5.66(1 \mathrm{H}$, $\mathrm{brs}, \mathrm{OH}), 6.48(1 \mathrm{H}, \mathrm{d}, J 8,17-\mathrm{H}), 6.63(1 \mathrm{H}, \mathrm{t}, J 8,15-\mathrm{H}), 6.74(1$
$\mathrm{H}, \mathrm{d}, J 8,14-\mathrm{H})$ and $7.14(1 \mathrm{H}, \mathrm{t}, J 8,16-\mathrm{H}) ; v_{\max }\left(\mathrm{CHCl}_{3}\right) / \mathrm{cm}^{-1}$ $3500-3100(\mathrm{OH}) 1645,1605$ and $1555 ; \lambda_{\max } / \mathrm{nm} 257\left(\varepsilon / \mathrm{dm}^{3}\right.$ $\mathrm{mol}^{-1} \mathrm{~cm}^{-1} 12340$ ) and 308 (19640); $m / z(\mathrm{EI}) 324\left(\mathrm{M}^{+}, 91 \%\right)$, 280 (90) and 77 (100) (Found: $\mathrm{M}^{+} 324.1825 . \mathrm{C}_{20} \mathrm{H}_{24} \mathrm{~N}_{2} \mathrm{O}_{4}$ requires $M, 324.1838$ ).

Further elution with $2.5 \%$ methanol in chloroform gave the more polar alcohol 37 or $36(15.7 \mathrm{mg}, 37 \%)$ as a colourless solid; m.p. $206-207^{\circ} \mathrm{C}$; $\delta_{\mathrm{H}} 1.37\left(3 \mathrm{H}, \mathrm{d}, J 7, \mathrm{CH}_{3} \mathrm{CHOH}\right)$, 1.8-2.5 ( $6 \mathrm{H}, \mathrm{m}$, methylenes), 2.30 ( $1 \mathrm{H}, \mathrm{dd}, J 6.5,3,3-\mathrm{H}$ ), 2.92 $\left(3 \mathrm{H}, \mathrm{s}, \mathrm{NCH}_{3}\right), 3.2-3.8\left(4 \mathrm{H}, \mathrm{m}, 8-\mathrm{H}\right.$ and $\left.10-\mathrm{H}_{2}\right), 3.46(1 \mathrm{H}$, $\mathrm{br} \mathrm{s}, \mathrm{OH}), 3.86(1 \mathrm{H}, \mathrm{d}, J 6.5,2-\mathrm{H}), 4.12(1 \mathrm{H}, \mathrm{m}$, $\left.\mathrm{CH}_{3} \mathrm{CHOH}\right), 6.48(1 \mathrm{H}, \mathrm{d}, J 8,17-\mathrm{H}), 6.64(1 \mathrm{H}, \mathrm{t}, J 8,15-$ $\mathrm{H}), 6.80(1 \mathrm{H}, \mathrm{d}, J 8,14-\mathrm{H})$ and $7.14(1 \mathrm{H}, \mathrm{t}, J 8,16-\mathrm{H})$; $v_{\text {max }}\left(\mathrm{CHCl}_{3}\right) / \mathrm{cm}^{-1} 3600-3100,1645,1605$ and $1555 ; \lambda_{\text {max }} / \mathrm{nm}$ $260\left(\varepsilon / \mathrm{dm}^{3} \mathrm{~mol}^{-1} \mathrm{~cm}^{-1} 13660\right)$ and 308 (22 453); $m / z(\mathrm{EI}) 324$ $\left(\mathrm{M}^{+}, 13 \%\right), 280(40), 157(94), 144$ (96) and 43 (100) (Found: $\mathrm{M}^{+}$324.1820. $\mathrm{C}_{20} \mathrm{H}_{24} \mathrm{~N}_{2} \mathrm{O}_{2}$ requires $M, 324.1838$ ).

Attempted Cyclisation of the Dione 29 with Boron TrifluorideDiethyl Ether.-The above dione 29 ( $50 \mathrm{mg}, 0.14 \mathrm{mmol}$ ) was dissolved in boron trifluoride-diethyl ether ( $5 \mathrm{~cm}^{3}$, freshly distilled) at $0^{\circ} \mathrm{C}$ under an atmosphere of nitrogen. After 30 min at $0^{\circ} \mathrm{C}$ the mixture was warmed to $20^{\circ} \mathrm{C}$ for 90 min by which time no starting material remained (TLC). The mixture was treated with an excess of dilute aqueous sodium carbonate ( 50 $\mathrm{cm}^{3}$ ) and extracted with diethyl ether ( $3 \times 15 \mathrm{~cm}^{3}$ ). The ether layers were combined, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and then the solvent was removed under reduced pressure to give a brown oil which was purified by flash chromatography. Elution with ethyl acetate gave the boron complex 38 as a colourless crystalline solid ( 22 $\mathrm{mg}, 38 \%$ ); m.p. 107-108 ${ }^{\circ} \mathrm{C}$ (diethyl ether-hexane); $\delta_{\mathrm{H}} 1.68(2 \mathrm{H}$, $\mathrm{m}, 5-\mathrm{H}_{2}$ ), 2.28 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3} \mathrm{CO}$ ), $2.30\left(2 \mathrm{H}, \mathrm{t}, J 7,4-\mathrm{H}_{2}\right.$ ), 3.15 ( 2 $\mathrm{H}, \mathrm{t}, J 7$, Ind- $\mathrm{CH}_{2}$ ), $3.18\left(2 \mathrm{H}, \mathrm{t}, J 6,6-\mathrm{H}_{2}\right), 3.43(2 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{COCH}_{2} \mathrm{CO}\right), 3.75\left(3 \mathrm{H}, \mathrm{s}, \mathrm{NCH}_{3}\right), 3.82(2 \mathrm{H}, \mathrm{t}, J 7$, Ind$\mathrm{CH}_{2} \mathrm{CH}_{2}$ ), $6.96(1 \mathrm{H}, \mathrm{s}$, Ind-2-H), $7.12(1 \mathrm{H}, \mathrm{t}, J 8$, Ind-5-H), $7.24(1 \mathrm{H}, \mathrm{t}, J 8$, Ind-6-H), $7.31(1 \mathrm{H}, \mathrm{d}, J$ 8, Ind-7-H) and 7.57 (1 $\mathrm{H}, \mathrm{d}, J 8$, Ind-4-H); $v_{\text {max }}\left(\mathrm{CHCl}_{3}\right) / \mathrm{cm}^{-1} 1720$ and $1620 ; \lambda_{\text {max }} / \mathrm{nm}$ $225\left(\varepsilon / \mathrm{dm}^{3} \mathrm{~mol}^{-1} \mathrm{~cm}^{-1} 27400\right)$ and 283 ( 16 180); $m / z(\mathrm{EI}) 388$ $\left(\mathrm{M}^{+}, 9 \%\right), 340$ (2), 157 (67) and 144 (100) (Found: $\mathrm{M}^{+}$ 388.1754. $\mathrm{C}_{20} \mathrm{H}_{23} \mathrm{~N}_{2} \mathrm{O}_{3} \mathrm{BF}_{2}$ requires $M, 388.1759$ ).
( $2 \mathrm{~S}^{*}, 3 \mathrm{~S}^{*}, 12 \mathrm{R}^{*}$ )-5-Deethyl-5,19-didehydro-3-methoxycarb-onyl-1-methyl-4-oxoaspidospermidine 39.-To a stirred solution of the keto ester 30 ( $62 \mathrm{mg}, 0.175 \mathrm{mmol}$ ) in dichloromethane ( 5 $\mathrm{cm}^{3}$ ) at $20^{\circ} \mathrm{C}$ under an atmosphere of nitrogen was added TFAA ( $0.25 \mathrm{~cm}^{3}, 1.75 \mathrm{mmol}$ ). The solution was stirred for 24 h at $20^{\circ} \mathrm{C}$ by which time no starting material remained. The mixture was added to an excess of saturated aqueous sodium hydrogen carbonate and extracted with dichloromethane ( $3 \times 10 \mathrm{~cm}^{3}$ ). The combined extracts were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and the solvent was removed under reduced pressure to give a yellow oil. Flash chromatography and elution with ethyl acetate gave the title didehydroaspidospermidine 39 as a colourless solid ( $30 \mathrm{mg}, 51 \%$ ); m.p. $204-206^{\circ} \mathrm{C}$ (diethyl ether); $\delta_{\mathrm{H}} 1.8-2.4$ $\left(6 \mathrm{H}, \mathrm{m}, 3 \times \mathrm{CH}_{2}\right), 2.74\left(3 \mathrm{H}, \mathrm{s}, \mathrm{NCH}_{3}\right), 3.24(1 \mathrm{H}, \mathrm{d}, J 8,3-\mathrm{H})$, $3.28(1 \mathrm{H}, \mathrm{m}, 8-\mathrm{H}$ or $10-\mathrm{H}), 3.38(1 \mathrm{H}, \mathrm{t}, J 10,8-\mathrm{H}$ or $10-\mathrm{H})$, $3.50(1 \mathrm{H}, \mathrm{dt}, J 13,4,8-\mathrm{H}$ or $10-\mathrm{H}), 3.74(1 \mathrm{H}$, ddd, $J 12,10,6$, $8-\mathrm{H}$ or $10-\mathrm{H}), 3.78\left(3 \mathrm{H}, \mathrm{s}, \mathrm{OCH}_{3}\right), 4.32(1 \mathrm{H}, \mathrm{d}, J 8,2-\mathrm{H}), 6.52$ ( $1 \mathrm{H}, \mathrm{d}, J 8,17-\mathrm{H}), 6.71(1 \mathrm{H}, \mathrm{t}, J 8,15-\mathrm{H}), 6.89(1 \mathrm{H}, \mathrm{d}, J 8$, $14-\mathrm{H})$ and $7.18(1 \mathrm{H}, \mathrm{t}, \mathrm{J}, 16-\mathrm{H}) ; v_{\text {max }}\left(\mathrm{CHCl}_{3}\right) / \mathrm{cm}^{-1} 1735,1645$ and 1575; $\lambda_{\text {max }} / \mathrm{nm} 253\left(\varepsilon / \mathrm{dm}^{3} \mathrm{~mol}^{-1} \mathrm{~cm}^{-1} 12440\right)$ and 308 (21530); m/z (EI) 338 ( $\mathrm{M}^{+}, 11 \%$ ) 279 (10), 260 (20), 229 (25), 157 (85), 151 (100) and 144 (75) (Found: $\mathrm{M}^{+} 338.1648$. $\mathrm{C}_{20} \mathrm{H}_{22} \mathrm{~N}_{2} \mathrm{O}_{3}$ requires $M, 338.1630$ ).

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[^0]:    * One enantiomer only is shown in structural formulae of racemates.

